

Determination of Bismuth in Cosmetics

Bismuth compounds are often used in cosmetics as a replacement for lead given its toxicity. Bismuth can be measure by hydride generation atomic fluorescence spectrometer. The sample preparation consists of oxidizing Bi to Bi³⁺, which reacted with NaBH₄ yields BiH₃. The BiH₃ is moved to the atomizer by the courier gas and broken down there. The ground state of Bi can be excited by the light by the Bi-HC lamp and the fluorescence sign is measured consequently by photo multiple tube.

1. Major equipment and reagents

- LUMINA 3300 atomic fluorescence spectrometer with Se lamp
- NaBH₄ solution:
- Dissolve 10 g NaBH₄ in 500 mL (1 g/L NaOH) solution

Bismuth standard solution:

- Commercial available selenium standard solution (1000 ppm), diluted to 1 ppm
- Thiourea solution:
- Measure 0.7 g thiourea, dissolve in 100 mL distilled water

$Mg(NO_3)_2$ solution:

- Measure 500 g Mg(NO₃)₂ $6H_2O$, add distilled to the total volume at 1000 mL
- MgO
- Concentrated HNO₃ (65~68%)
- Concentrated HCI (36~38%)
- Concentrated HClO
- Concentrated H₂SO
- High pure argon (>99.99%)
- High pure distilled water

2. Method

Measure 1 g sample into a 150 mL Erlenmeyer flask, add 10 mL HNO₃ and 5 mL H₂O₂ Place on hot plate and heat until no yellowish smoke comes out, cool down, add 3 mL HClO₄, $5m_L H_2SO_4$, and heat on plate again. If the color of the mixture is brown, add more HNO₃ and repeat heating step until the solution becomes clear. Cool down and add 20 mL of distilled water, heat until white smoke comes out and add water again. Repeat this procedure three times and remove the remaining to the volumetric flask and add water to a final volume of 50 mL.

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3. Instrument parameters

Carrier gas	500 mL/min
Shield gas	700 mL/min
HCL current	100 mA
PMT voltage	400 V
Integration time	6 s
Pump speed	50 r/min
Reducing reagent solution	2% NaBH₄ in 0.1% NaOH

4. Results

This method gives:

Detection limit: 0.7ppb, Recovery rate: 90~118% Relative standard deviation: 2~7%



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